ENANTIOSELECTIVE TOTAL SYNTHESIS OF (+)-a-2,7,11-CEMBRATRIENE-4,6-DIOL (a-CBT)

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Summary: The total synthesis of (+)-a-CBT (16) was effected from the achiral 17-membered ketone 3 by a sequence featuring asymmetric reduction, diastereoselective [2,3] Wittig ring contraction and hydroxyl directed epoxidation as stereochemically important steps.

The diastereomeric tobacco cembranes a- and β -CBT¹ (I and II) have been shown to inhibit both plant growth² and tumor promotion.³ We recently reported a total synthesis of the racemic β -isomer II by a route featuring a novel diastereoselective [2,3] Wittig ring contraction of the 17-membered propargylic ether III.⁴ The present study was undertaken to develop a stereorational synthesis of a-CBT that would define both the relative and absolute stereochemistry of the natural product.



We have previously shown that a bulky alkoxyl substituent at C6 of the macrocyclic ether III exerts considerable conformational control on the [2,3] Wittig ring contraction leading to a marked preference for the *trans,syn* diastereoisomer (cf. $7 \rightarrow 8$).⁴ It was of interest to prepare this ether in a configurationally defined manner and employ the resulting scalemic⁵ rearrangement product 8 for the synthesis of natural a-CBT or its enantiomer.

In line with configurational assignments established for other tobacco derived cembranoids,⁶ we selected the R enantiomer for initial study. We had hoped to prepare the (R)-alcohol 5 by asymmetric reduction of the propargyl ketone IV, but all attempts at oxidation of the racemic alcohol (III, R = H) led to the conjugated allenic ketone instead. In order to circumvent the apparently facile alkyne-allene isomerization we turned to the Co complex 2, readily prepared by treatment of racemic alcohol 1 with $Co_2(CO)_8$ in ether.⁷ Oxidation by the Swern protocol⁸ afforded ketone 3 which was reduced to the Co-complexed homopropargylic alcohol 4 with the LiAlH₄-Darvon alcohol ("Chirald[®]") complex.⁹ Treatment of 4 with ceric ammonium nitrate led to the desired alcohol 5 of 71% ee whose configuration was assigned through ¹H NMR analysis of the (R)-O-methyl mandelate 6 as previously described for related compounds.¹⁰



(a) $Co_2(CO)_8$, Et_2O (98%); (b) $(COCl)_2$, DMSO, Et_3N , CH_2Cl_2 (83%); (c) Chirald-LiAlH4, Et_2O , -78°C; (d) $(NH_4)_2Ce(NO_3)_6$, MeOH, Et_2O , (90%, two steps); (e) TBSCl, Et_3N , DMAP, CH₂Cl₂ (89%); (f) n-BuLi-TMEDA, THF, pentane, -78°C (85%); (g) Me₂CuLi, Et_2O , 0°C (90%); (h) DIBAH, THF, -78°C (96%); (i) (Ph₃P)₃RhCl, H₂, C₆H₆, EtOH (69%); (j) VO(acac)₂, t-BuOOH, PhCH₃ (88%); (k) CH₃SO₂Cl, C₅H₅N, CH₂Cl₂, 0°C (92%); (l) Bu₄NF, THF (92%); (m) Na, NH₃, THF, -40°C (75%)

[2,3] Wittig ring contraction of the TBS ether 7 was effected with n-BuLi • TMEDA affording a 73:11:16 mixture of the *trans,syn:trans,anti:cis,syn* diastereoisomers¹¹ favoring the *trans,syn* isomer 8.4 Swern oxidation⁸ of this mixture gave ketone 10 as a 90:10 mixture of syn and anti isomers. Addition of Me₂CuLi yielded the (E)-enone 11 also as a 90:10 syn:anti mixture plus a small amount (<10%) of the isomeric (Z)-isomer(s).⁴ This mixture was reduced with (*i*-Bu)₂AlH and selectively hydrogenated with Wilkinson's catalyst¹² affording alcohol 12 in 69% yield after separation from isomeric impurities by column chromatography.

Epoxidation of allylic alcohol 12 with VO(acac)₂, t-BuOOH afforded a single epoxide 13. The epoxide stereochemistry was confirmed through conversion to the keto epoxide 17 which was isomeric with the diastereomer 18 derived from 19, a substance whose structure was previously determined, as the racemate, by x-ray analysis.⁴



(a) (COCl)₂, DMSO, Et₃N, CH₂Cl₂; (b) K₂CO₃, MeOH; TBSCl, Et₃N, DMAP, CH₂Cl₂; (Ph₃P)₃RhCl, C₆H₆, EtOH, H₂

The final steps of the sequence were effected along lines of our β -CBT synthesis.⁴ Thus epoxy alcohol 13 was converted to epoxy mesylate 15 through treatment with methanesulfonyl chloride in pyridine and desilylation. The observed H2/H3 coupling of 9.6 Hz is suggestive of a favored anti arrangement for this epoxy mesylate. Reductive elimination with sodium ammonia afforded the (E) allylic diol 16, $[a]_D + 90^\circ$, identical according to ¹H, ¹³C NMR and TLC comparison with an authentic sample. The ee of this material was calculated as 73% by ¹⁹F NMR analysis of the Mosher ester derivative. Since the rotation of the synthetic and natural material is of the same sign and comparable magnitude, correcting for ee, the absolute configuration of a-CBT can be assigned as 15,4S,6R (16).

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